

# Deamidation alters interactions of $\beta$ -crystallins in hetero-oligomers

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**Purpose:** Cataracts are a major cause of blindness worldwide. A potential mechanism for loss of visual acuity may be due to light scattering from disruption of normal protein–protein interactions. During aging, the lens accumulates extensively deamidated crystallins. We have previously reported that deamidation in the  $\beta$ A3-crystallin ( $\beta$ A3) dimer decreased the stability of the dimer in vitro. The purpose of the present study was to investigate if deamidation altered the interaction of  $\beta$ A3 with other  $\beta$ -crystallin subunits.

**Methods:** Deamidation was mimicked by replacing glutamines, Q85 and Q180, at the predicted interacting interface between  $\beta$ A3 domains with glutamic acids by site-directed mutagenesis. Human recombinant wild type  $\beta$ A3 or the doubly deamidated mutant  $\beta$ A3 Q85E/Q180E (DM  $\beta$ A3) were mixed with either  $\beta$ B1- or  $\beta$ B2-crystallin ( $\beta$ B1 or  $\beta$ B2) subunits. After incubation at increasing temperatures, hetero-oligomers were resolved from individual subunits and their molar masses determined by size exclusion chromatography with in line multiangle laser light scattering. Structural changes of hetero-oligomers were analyzed with fluorescence spectroscopy and blue-native PAGE.

**Results:** Molar masses of the hetero-oligomer complexes indicated  $\beta$ A3 formed a polydispersed hetero-tetramer with  $\beta$ B1 and a monodispersed hetero-dimer with  $\beta$ B2. Deamidation at the interface in the  $\beta$ A3 dimer decreased formation of the hetero-oligomer with  $\beta$ B1 and further decreased formation of the hetero-dimer with  $\beta$ B2. During thermal-induced denaturation of the deamidated  $\beta$ A3 dimer,  $\beta$ B1 but not  $\beta$ B2 was able to prevent precipitation of  $\beta$ A3.

**Conclusions:** Deamidation decreased formation of hetero-oligomers between  $\beta$ -crystallin subunits. An excess accumulation of deamidated  $\beta$ -crystallins in vivo may disrupt normal protein–protein interactions and diminish the stabilizing effects between them, thus, contributing to the accumulation of insoluble  $\beta$ -crystallins during aging and cataracts.

During normal aging and cataract formation, loss of visual acuity is accompanied by extensive modification of lens proteins. Numerous types of modifications have been identified and include truncation, methylation, oxidation, disulfide bond formation, advanced glycation end products, and deamidation [1-8]. These modifications change the structure and stability of lens proteins [9,10].

In the intact lens, the major structural proteins, called crystallins, are tightly packed in a very high concentration of 300–500 mg/ml [11]. However, upon processing of the lens, crystallins separate into soluble and insoluble fractions, with the insoluble proteins increasing during aging [12,13]. Modifications increase in the insoluble proteins [8,14], suggesting a potential mechanism for compromised lens function.

Of the potential post-translational modifications, deamidation is one of the major modifications in the lens and is significantly increased in aged and cataractous lenses [8]. Deamidation sites have been identified in every crystallin, and are particularly prevalent in  $\beta$ -crystallins [7,8,14]. For example, at least nine deamidation sites have been identified

in  $\beta$ A3-crystallin ( $\beta$ A3) [8]. Furthermore, several deamidation sites are greater in the insoluble than in the soluble proteins [8,15], strongly suggesting an association between deamidation and insolubilization. The exact cause for insolubilization is not known, but may involve disruption of the normal interactions between  $\beta$ -crystallins.

The  $\beta$ -crystallins are comprised of seven polypeptide chains,  $\beta$ A1-,  $\beta$ A2-,  $\beta$ A3-,  $\beta$ A4-,  $\beta$ B1-,  $\beta$ B2-, and  $\beta$ B3-crystallin that when isolated in vitro form complex hetero-oligomers of dimers,  $\beta$ Low- ( $\beta$ L), and higher-ordered oligomers,  $\beta$ High-crystallin ( $\beta$ H) [2,3,16]. During normal maturation and aging, a hetero-oligomer intermediate in size between  $\beta$ L and  $\beta$ H increases [2] and is associated with post-translational modifications.

Two different  $\beta$ -crystallin dimer structures have been resolved by crystallography [17,18]. Dimers are comprised of two domains connected by a short linker peptide that is either bent as in  $\beta$ B1-crystallin ( $\beta$ B1) or extended as in  $\beta$ B2-crystallin ( $\beta$ B2). Each domain has two Greek key motifs comprised of  $\beta$ -strands. This interface has been well characterized, particularly in  $\beta$ B2, and hydrophobic interactions with salt bridges are thought to stabilize the interacting interface of the dimer [19-21]. Similar interactions most likely occur between different  $\beta$ -subunits.  $\beta$ B1 and  $\beta$ B2 have been reported to make hetero-oligomers with  $\beta$ A3 in vitro [22,23].

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We have previously reported the effects of deamidation at the critical interacting interface in  $\beta$ -crystallins [10,24-28]. Stability was significantly decreased due to deamidation, without the homo-dimer dissociating. Deamidation introduces a negative charge at physiologic pH and most likely created a cavity or space at the interface that contributed to the structural changes that decreased stability [21]. The purpose of this study was to investigate the effect of deamidation at the interface in  $\beta$ A3 on subunit-subunit interactions with other  $\beta$ -subunits to mimic the more complex interactions found *in vivo*.

## METHODS

**Expression and purification of recombinant protein:** Wild type (WT) human  $\beta$ -crystallins,  $\beta$ B1,  $\beta$ B2, and  $\beta$ A3, were recombinantly expressed in *E. coli* as described previously [24,26,27]. Deamidation was mimicked by replacing glutamines with glutamic acids, using site directed mutagenesis to generate the Q85E/Q180E  $\beta$ A3 (DM  $\beta$ A3) mutant as previously reported [27] (Quick Change Mutagenesis, Stratagene, Cedar Creek, TX).

All human recombinant  $\beta$ -crystallins were purified by successive ion-exchange chromatography [10,27,28] and the purity of the proteins was checked by SDS-PAGE and mass

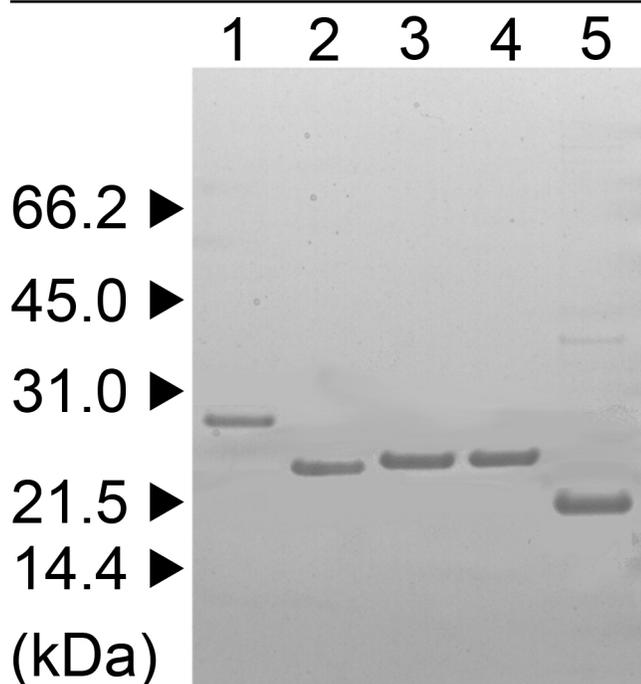


Figure 1. Purified recombinant crystallins. Recombinantly expressed proteins were purified to near homogeneity as indicated by a single band on SDS-PAGE. Each protein (1  $\mu$ g) was visualized with Coomassie blue stain on a 1.0 mm thick, 4-12% Bis/Tris gel. Proteins were  $\beta$ B1 (lane 1),  $\beta$ B2 (lane 2),  $\beta$ A3 WT (lane 3),  $\beta$ A3 DM (lane 4) and  $\alpha$ A-crystallin (lane 5). Triangles indicate positions of molecular weight markers.

spectrometry. Proteins were flash frozen in liquid nitrogen and stored at  $-80^{\circ}\text{C}$  stock without lyophilization. Human  $\alpha$ A-crystallin was expressed and purified as previously described [29].

**Size exclusion chromatography with in line multi angle laser light scattering (SEC-MALS):** To investigate the effects of deamidation on the formation of hetero-oligomers as a function of temperature, a one to one molar ratio of recombinant WT  $\beta$ A3 or DM  $\beta$ A3 was premixed with either  $\beta$ B1 or  $\beta$ B2 at a final concentration of 0.4–0.5 mg/ml followed by incubation for 90 min at  $37^{\circ}\text{C}$  or  $55^{\circ}\text{C}$ . Samples were equilibrated in buffer (pH 6.8) containing 29 mM  $\text{Na}_2\text{HPO}_4$ , 29 mM  $\text{NaH}_2\text{PO}_4$ , 100 mM KCl, 1 mM EDTA, and 1 mM DTT. The mixture was filtered and injected onto a SEC-MALS system with in-line refractive index detector (HELEOS and Opti-Lab instruments; Wyatt Technology Inc. Santa Barbara, CA). Molar masses were calculated from the intensity of scattered light at 18 different angles according to Rayleigh light scattering principles and using software provided by the manufacturer. The protein concentration was determined from the refractive index or UV detector and a  $dn/dc$  value of 0.185 was used [30].

To investigate the effects of deamidation on the formation of hetero-oligomers as a function of time at physiologic

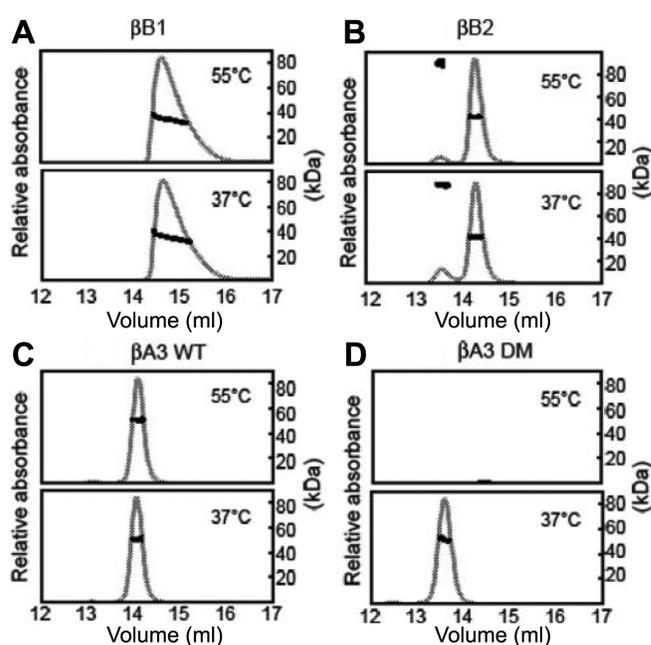


Figure 2. Homo-oligomerization of  $\beta$ -crystallins. Purified crystallins were incubated for 90 min at either  $37^{\circ}\text{C}$  or  $55^{\circ}\text{C}$  and then subjected to SEC-MALS. Eluting proteins were detected by UV absorption at 280 nm (grey line). Molar masses were determined across the peak (solid squares). From molar masses,  $\beta$ B1 was a mixture of monomer-dimers (A);  $\beta$ B2 was predominantly a dimer with less of an earlier eluting peak (B);  $\beta$ A3 WT was a dimer (C); and  $\beta$ A3 DM was a dimer at  $37^{\circ}\text{C}$ , but had precipitated at  $55^{\circ}\text{C}$  (D).

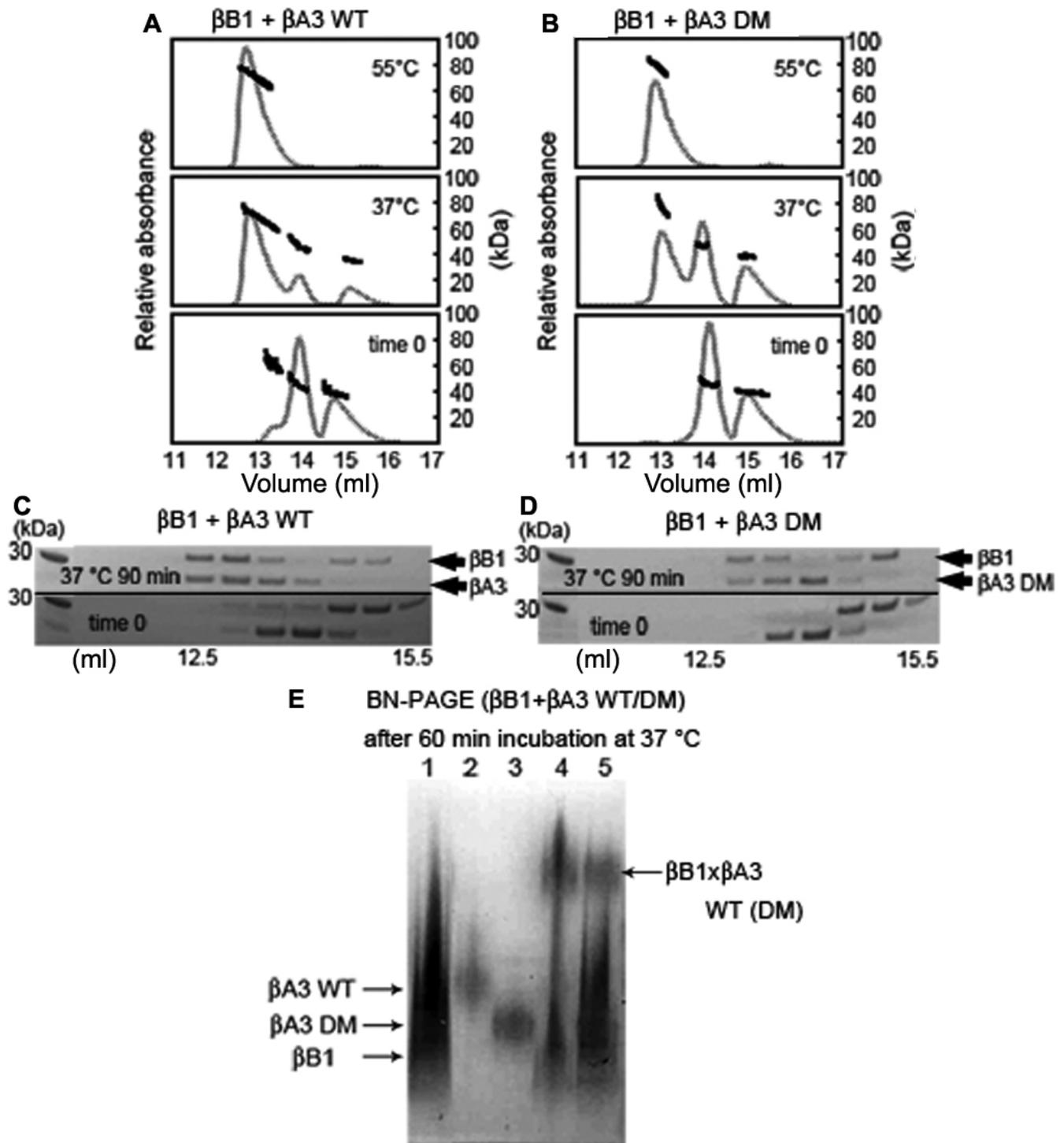


Figure 3. Hetero-oligomerization of  $\beta B1$  with  $\beta A3$ . Similar size oligomers were obtained when  $\beta B1$  was mixed with either  $\beta A3$  WT (A) or  $\beta A3$  DM (B). Mixtures were immediately subjected to SEC-MALS (bottom panel) or incubated at 37 °C (middle panel) or 55 °C (top panel) for 90 min as in Figure 2. Note the decreased amount of the  $\beta B1:\beta A3$  DM compared to the  $\beta B1:\beta A3$  WT complex. Eluting fractions of  $\beta B1$  mixed with either  $\beta A3$  WT (C) or  $\beta A3$  DM (D) were collected and subjected to SDS-PAGE. Both subunits,  $\beta B1$  and  $\beta A3$  WT or DM, were present in the complex. Hetero-oligomer formation was confirmed by subjecting mixtures of  $\beta B1$  with  $\beta A3$  WT or  $\beta A3$  DM incubated at 37 °C for 60 min to Blue-Native-PAGE (E). Proteins were  $\beta B1$  (lane 1),  $\beta A3$  WT (lane 2),  $\beta A3$  DM (lane 3),  $\beta B1:\beta A3$  WT (lane 4), and  $\beta B1:\beta A3$  DM (lane 5).

temperature, mixtures prepared as described above were incubated for 30, 60, 90, 120 and 300 min at 37 °C. Additionally, a mixture of a 5:1 molar ratio of  $\beta$ B1 to  $\beta$ A3 was incubated for 30 min, to assure a concentration favoring dimer formation of  $\beta$ B1 [10]. Concentrations of individual proteins were calculated using the UV absorbance at 280 nm and extinction coefficients of  $2.07 \text{ (mg/ml)}^{-1}\text{cm}^{-1}$  for  $\beta$ B1,  $1.71 \text{ (mg/ml)}^{-1}\text{cm}^{-1}$  for  $\beta$ B2, and  $2.62 \text{ (mg/ml)}^{-1}\text{cm}^{-1}$  for  $\beta$ A3. Samples (100  $\mu$ l) were injected onto a Superose 12 10/300 GL column (Amersham Biosciences Corp., Piscataway, NJ) equilibrated in the same buffer with a flow rate of 0.2 ml/min.

**Circular dichroism and fluorescence of hetero-oligomers:** The eluted peaks from SEC-MALS containing hetero-oligomers were pooled and concentrated to 1.0 mg/ml, then exhaustively dialyzed into 5 mM  $\text{NaH}_2\text{PO}_4$  and 5 mM  $\text{Na}_2\text{HPO}_4$  (pH 6.8), containing 100 mM NaF for analysis by circular dichroism (CD). Measurements were obtained using a JASCO J-810 spectropolarimeter (JASCO, Easton, MD). Hetero-oligomer samples were measured at 0.15 mg/ml in a 0.1 cm cell for far-UV scans. Experiments were repeated on two different protein preparations. Concentrations of hetero-oligomers were determined by the BCA assay (Pierce, Rockford, IL) and by amino acid analysis (Molecular Structure Facility, UC Davis, Davis, CA).

The tertiary structures of hetero-oligomers were determined by fluorescence spectrometry. The fluorescence buffer (pH 7.0) contained 50 mM  $\text{Na}_2\text{HPO}_4$ , 50 mM  $\text{NaH}_2\text{PO}_4$ , 5 mM DTT, and 2 mM EDTA. Proteins at 1  $\mu$ M were incubated for 24 h at 22 °C, and measured on a Photon Technology International QM-2000-7 spectrometer using the

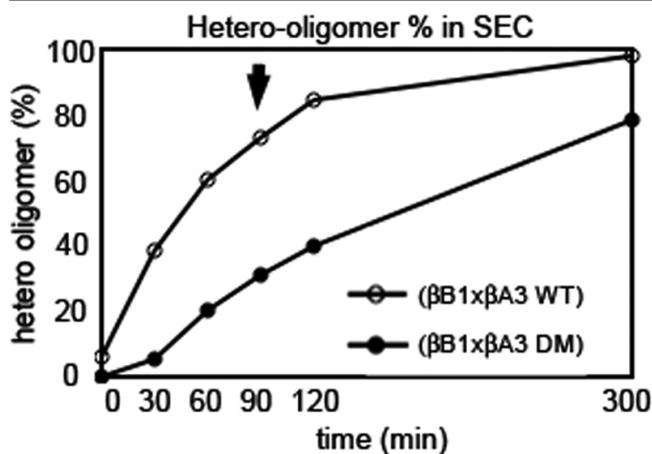


Figure 4. Hetero-oligomerization of  $\beta$ B1 with  $\beta$ A3 as a function of time.  $\beta$ B1 was mixed with either  $\beta$ A3 WT (open circles) or  $\beta$ A3 DM (closed circles) and incubated at 37 °C from 0-300 min and then subjected to SEC-MALS. The relative amount of complex was determined from the peak area. Formation of the  $\beta$ B1: $\beta$ A3 DM complex was slower than formation of the  $\beta$ B1: $\beta$ A3 WT complex. Arrow indicates 90 min time point showing 40 % difference between WT and DM.

manufacturer's supplied software, FeliX (Photon Technology International, Lawrenceville, NJ). Emission spectra were recorded between 300 and 400 nm with an excitation wavelength at 283 nm or 293 nm. Slit widths were set to 2 nm. Emission spectra were corrected for the buffer signal.

**Shape determination by blue-native-electrophoresis (BN-PAGE):** To investigate the shape of each hetero-oligomer, the same samples analyzed by SEC-MALS were analyzed by BN-PAGE. Samples were mixed in a 1:1 ratio (v/v) of sample buffer (141 mM Tris-HCl, 106 mM Tris-Base, 0.5 mM EDTA, 15% [v/v] glycerol, 0.002% Bromophenol blue at pH 7.0). Blue-native electrophoresis was performed as previously described with some modifications using pre-cast, 1.0 mm thick 8x8 cm, polyacrylamide Nu-PAGE 4%–12% Bis-Tris gels (Invitrogen, Carlsbad, CA). Anode buffer was 50 mM Bis-Tris (pH 7.0) and cathode buffer was 15 mM Bis-tris, 50 mM Tricine, 0.02% coomassie brilliant blue-G250 (pH 7.0). BN-PAGE was performed at 200 V and 4 °C until the blue dye reached the bottom of the gel. Proteins were visualized with SimplyBlue SafeStain (Invitrogen) after overnight washing with water.

**Heat induced precipitation:** Heat induced precipitation of complex  $\beta$ -crystallin was measured in a thermal jacketed cuvette with constant stirring (Cary 4 Bio UV-Visible spectrophotometer; Varian). Heat incubation was performed in the same buffer as the SEC-MALS experiments described

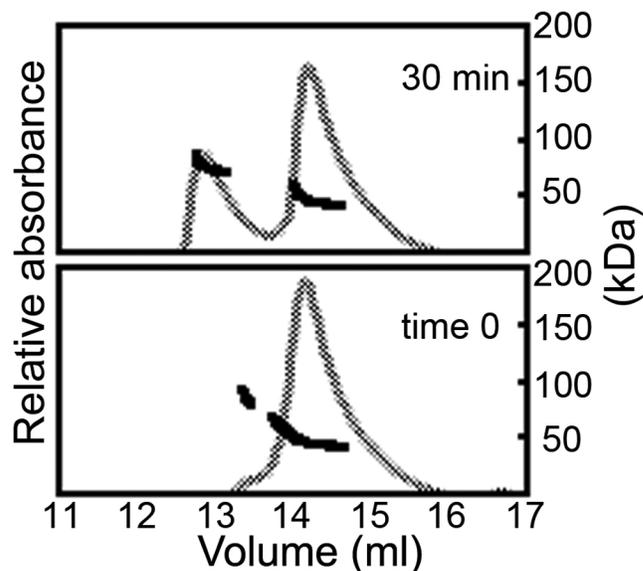


Figure 5. Hetero-oligomerization of  $\beta$ B1 dimers with  $\beta$ A3 dimers. In order to mix  $\beta$ B1 dimers with  $\beta$ A3 dimers, the concentration of  $\beta$ B1 was increased to favor  $\beta$ B1 dimer formation. A 5:1 molar ratio of  $\beta$ B1: $\beta$ A3 was mixed and immediately subjected to SEC-MALS (bottom panel) or after 30 min at 37 °C (top panel). Molar masses are indicated by the closed squares and the UV absorbance at 280 nm by the grey tracing. Similar molar masses are obtained for the hetero-oligomer as in Figure 3.

above. All samples at 1:1 molar ratios were concentrated to a final total concentration of 0.1 mg/ml. The turbidity of the proteins, as an indicator of aggregation and precipitation was monitored at 405 nm. All samples were added to preheated buffer in the cuvette and incubated at 55 °C for 180 min.

## RESULTS

*Homo-oligomerization of  $\beta$ -crystallins:* Recombinantly expressed crystallins were purified to greater than 95% purity indicated by a single band on SDS-PAGE (Figure 1).  $\beta$ B1,  $\beta$ B2,  $\beta$ A3, and DM  $\beta$ A3 were incubated for 90 min at 37 °C or 55 °C, and subjected to SEC-MALS (Figure 2A-D). All samples eluted in a single peak with  $\beta$ B2 having an additional earlier eluting peak.  $\beta$ B1 eluted as a monomer-dimer mixture.  $\beta$ B2 eluted mostly as a dimer with a lesser amount of a tetramer, and WT  $\beta$ A3 and DM  $\beta$ A3 eluted as dimers.  $\beta$ B2 and WT  $\beta$ A3 eluted as dimers at both temperatures. A DM  $\beta$ A3 dimer was not observed at 55 °C by SEC-MALS, due to precipitate forming that would have been filtered immediately before SEC-MALS (Figure 2D and [27]).

*Hetero-oligomerization of  $\beta$ A3 with  $\beta$ B1:* Both WT  $\beta$ A3 and DM  $\beta$ A3 formed a complex with  $\beta$ B1 that eluted as a polydispersed peak at 37 °C (Figure 3A,B). The protein composition of the hetero-oligomer peak visualized by SDS-PAGE indicated the presence of equal amounts of each subunit suggesting predominantly hetero-tetramer formation (Figure 3C,D). The appearance of diffused high molecular weight bands of the mixed samples resolved by BN-PAGE also indicated polydispersity (Figure 3E). Complex formation increased at 55 °C, accompanied by a decrease in the amount of individual  $\beta$ B1 and  $\beta$ A3 protein peaks.

Next,  $\beta$ B1 was mixed with WT  $\beta$ A3 or DM  $\beta$ A3 at 37 °C as a function of time (Figure 4). Immediately upon mixing  $\beta$ B1 with WT  $\beta$ A3, a higher-ordered peak was detected by SEC-MALS, not present with DM  $\beta$ A3. While, the total amount of the  $\beta$ B1-WT  $\beta$ A3 complex and the  $\beta$ B1-DM  $\beta$ A3 complex were nearly the same after 300 min incubation, the  $\beta$ B1-DM  $\beta$ A3 complex formation was slower (Figure 4). Approximately 40% less  $\beta$ B1-DM  $\beta$ A3 complex formed than the  $\beta$ B1-WT  $\beta$ A3 at 37 °C after 90 min.

$\beta$ B1 was a mixture of monomer and dimers at the protein concentrations used (Figure 2A). To mix a dimer  $\beta$ B1 with a dimer  $\beta$ A3, experiments were repeated with a 5:1 ratio of  $\beta$ B1 to  $\beta$ A3 (Figure 5). Both proteins were present as dimers as detected by the overlapping peaks in SEC-MALS at 0 min. After a 30 min incubation at 37 °C, a hetero-oligomer peak eluted at the same position as the 1:1 protein peak (Figure 5 and Figure 3A). This suggested that dimers associated to form the hetero-oligomers.

*Hetero-oligomerization of  $\beta$ A3 with  $\beta$ B2:* The  $\beta$ B2- $\beta$ A3 complex differed from that of the  $\beta$ B1- $\beta$ A3 complex and was predominantly a hetero-dimer at 37 °C.  $\beta$ B2 and  $\beta$ A3 subunits differed only slightly in their elution (Figure 2B,C), with the

shoulder present in Figure 6 attributed to  $\beta$ B2. After 90 min at 37 °C, there was a slight shift in the main peak partly masking the  $\beta$ B2 tetramer peak. The molar mass of 53-57 kDa at the peak suggested hetero-dimer formation (Figure 6A). After 90 min at 55 °C, there was a further shift in the WT  $\beta$ B2- $\beta$ A3 peak, with a molar mass from 67 to 73 kDa (Figure 6A) that suggested a higher-ordered oligomer similar to the  $\beta$ B1- $\beta$ A3 hetero-oligomer.

In contrast to WT  $\beta$ A3, DM  $\beta$ A3 did not readily form a complex with  $\beta$ B2 that was detected by SEC-MALS (Figure 6B). After 90 min at 37 °C, the shoulder disappeared with an increase in peak height. However, there was no shift in elution of the peak and the leading edge of the peak was at 55 kDa. After 90 min at 55 °C, there was also not a shift in elution of the peak and the late eluting  $\beta$ B2 shoulder was still present. Of note was the higher-ordered oligomer peak present immediately upon mixing  $\beta$ B2 and DM  $\beta$ A3 (Figure 6B). This peak did not increase with time or temperature and may represent a subpopulation of readily formed aggregate. Upon heating at 55 °C, there was a 50% decrease in the height of the main peak and precipitate was visible in the sample.

Because of the close molecular weights of  $\beta$ B2 and  $\beta$ A3, proteins were analyzed by BN-PAGE instead of SDS-PAGE (Figure 6C). WT  $\beta$ A3 and DM  $\beta$ A3 also migrated differently during BN-PAGE and suggested the two proteins had different shapes. After 60 min preincubation, samples were resolved by BN-PAGE for a minimum of 4 h at 4 °C. Diffused protein bands were visualized between  $\beta$ B2 and  $\beta$ A3 indicating a hetero-oligomer had formed, with less of the  $\beta$ B2-DM  $\beta$ A3 forming (Figure 6C). During the long analysis time for BN-PAGE, proteins may have formed complexes not detected during SEC-MALS.

*Effects of deamidation on the hetero-oligomer structure:* Circular dichroism measurements indicated different secondary structures for  $\beta$ B1 and  $\beta$ A3, with little difference between WT  $\beta$ A3 and DM  $\beta$ A3 as previously shown (Figure 7 and [10,27]). Circular dichroism spectra were also similar for complexes of either  $\beta$ B1 and WT  $\beta$ A3 or  $\beta$ B1 and DM  $\beta$ A3. Both hetero-oligomer complexes displayed secondary structure with maxima at 192 nm and minima at 204 nm. A slight minima at 235 nm was also observed, characteristic of  $\beta$ A3 [22]. The minima at 204 reflected the contribution of  $\beta$ B1 [10].

$\beta$ A3 has nine tryptophans with five buried and  $\beta$ B1 has eight tryptophans with four buried, resulting in similar fluorescence spectra (Figure 7 and [22]). Upon hetero-oligomer formation there was a slight blue-shift in the fluorescence peak to 335 nm (Figure 7). Emission spectra for these proteins were the same at either 283 nm or 293 nm. The higher-ordered structures were similar for the  $\beta$ B1-WT  $\beta$ A3 or  $\beta$ B1-DM  $\beta$ A3 hetero-oligomers.

*$\beta$ B1, but not  $\beta$ B2 prevented thermal aggregation of deamidated  $\beta$ A3:* To investigate the stability of  $\beta$ B1- $\beta$ A3

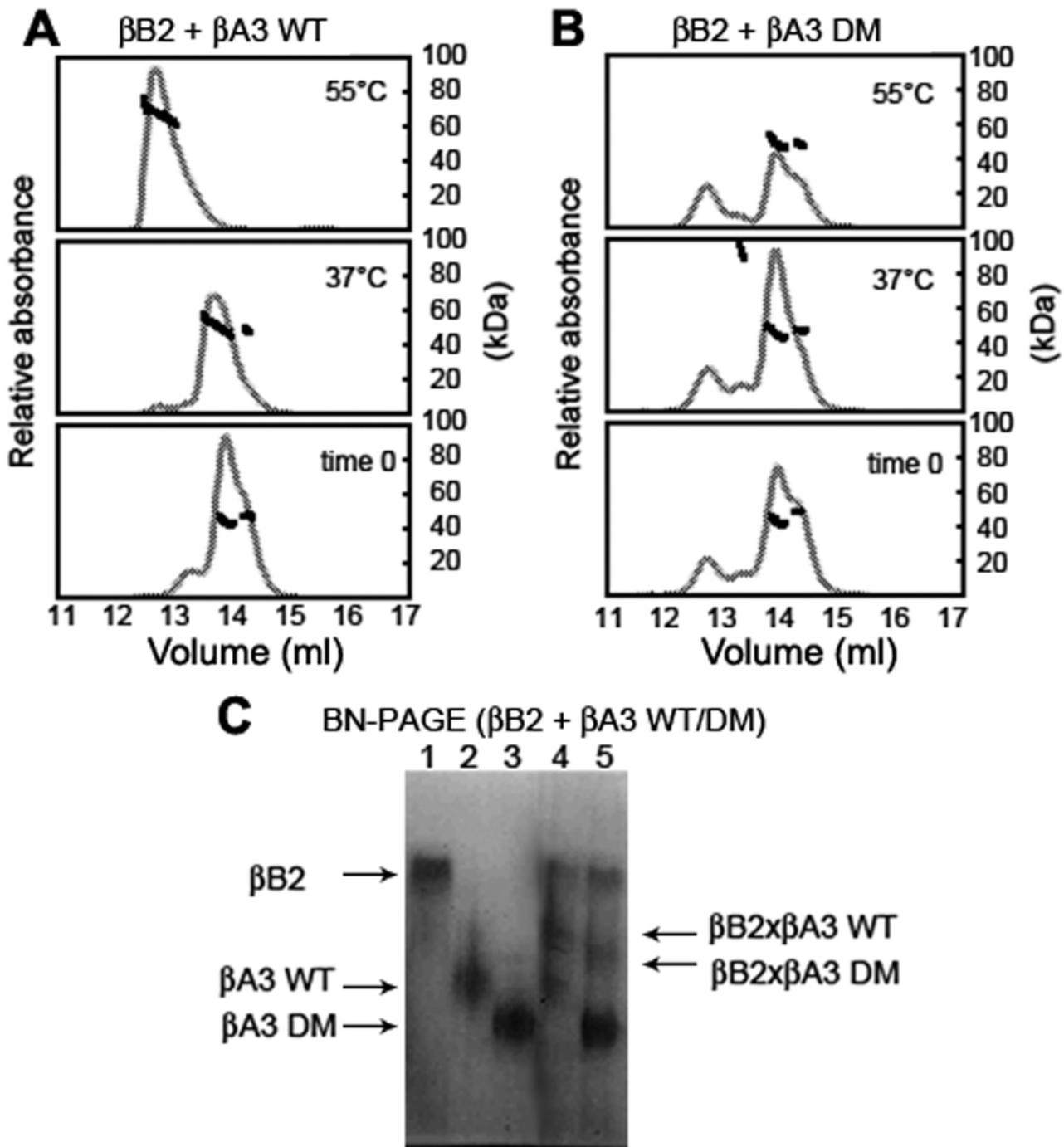


Figure 6. Hetero-oligomerization of  $\beta$ B2 with  $\beta$ A3.  $\beta$ B2 was mixed with either  $\beta$ A3 WT (A) or  $\beta$ A3 DM (B) and mixtures were immediately subjected to SEC-MALS (bottom panel) or incubated at 37 °C (middle panel) or 55 °C (top panel) for 90 min as for  $\beta$ B1 in Figure 3. Molar masses of  $\beta$ B2: $\beta$ A3 WT peaks indicated predominantly dimers at 37 °C with a slight shift in elution volume and oligomers at 55 °C. Molar masses of  $\beta$ B2: $\beta$ A3 DM peaks indicated predominantly dimers at 37 °C without a shift in elution volume and no oligomers were detected at 55 °C. Differences in hetero-oligomer formation were confirmed by subjecting mixtures of  $\beta$ B2 with  $\beta$ A3 WT or  $\beta$ A3 DM incubated at 37 °C for 60 min to Blue-Native-PAGE (C). Proteins were  $\beta$ B2 (lane 1),  $\beta$ A3 WT (lane 2),  $\beta$ A3 DM (lane 3),  $\beta$ B2: $\beta$ A3 WT (lane 4), and  $\beta$ B2: $\beta$ A3 DM (lane 5). Arrows indicate migration position of proteins.

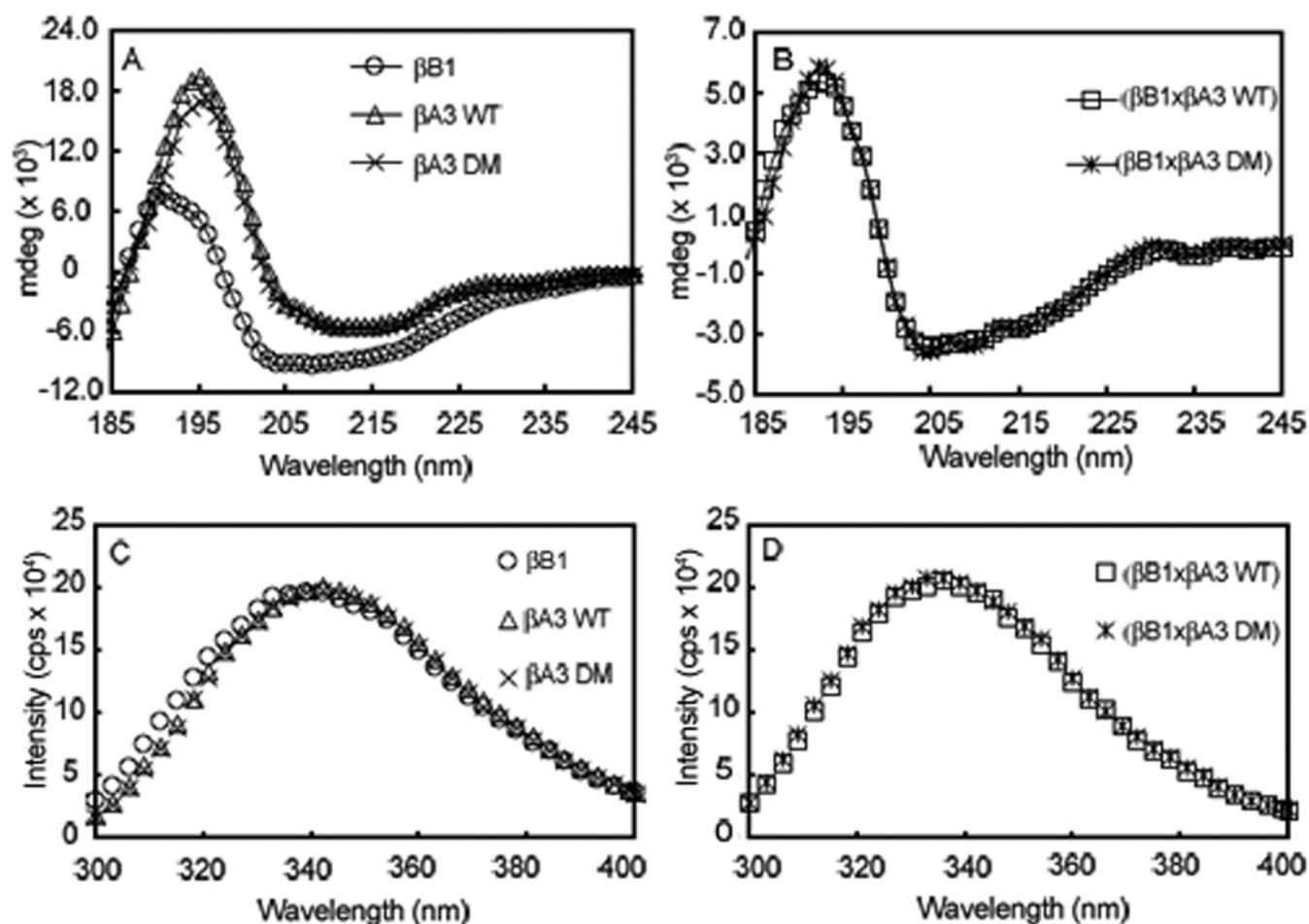


Figure 7. Structures of hetero-oligomer complexes. Eluting protein peaks from SEC-MALS were further analyzed by far-UV CD (A, B) and fluorescence spectrometry with excitation at 283 nm (C, D). Data were obtained of individual  $\beta$ -subunits (A, C) and complexes of  $\beta$ B1: $\beta$ A3 WT or DM (B, D) and indicated similar structure of the the  $\beta$ B1: $\beta$ A3 WT and  $\beta$ B1: $\beta$ A3 DM complexes.

hetero-oligomers, samples were subjected to heat denaturation. Both  $\beta$ B1 and  $\beta$ B2 prevented the precipitation of WT  $\beta$ A3 (Figure 8A). A major finding of the present study was that  $\beta$ B1, but not  $\beta$ B2 was able to prevent heat induced precipitation of DM  $\beta$ A3, similar to prevention by the chaperone,  $\alpha$ A-crystallin (Figure 8B). As a control, the model substrate, ALDH, aggregated during heating, which was prevented by  $\alpha$ A-crystallin, but not  $\beta$ B1 or  $\beta$ B2 crystallins (Figure 8C).

## DISCUSSION

The major findings of this study were: 1) deamidation in  $\beta$ A3 decreased its interaction with other  $\beta$ -crystallin subunits, 2) during thermal denaturation of deamidated  $\beta$ A3,  $\beta$ B1 was able to prevent precipitation, and 3) the predicted interface in the  $\beta$ A3 dimer may also be important in the  $\beta$ B1 hetero-oligomer. These results suggest that deamidation contributes to cataract formation by disrupting hetero-oligomer interactions.

*Deamidation decreases  $\beta$ A3 oligomerization with other  $\beta$ -crystallin subunits:* Deamidation at the predicted

interacting interface in the  $\beta$ A3-dimer decreased the hetero-oligomer formation of  $\beta$ A3 with either  $\beta$ B1 or  $\beta$ B2. This is significant because deamidation may disrupt higher-ordered oligomers necessary for lens transparency and contribute to the accumulation of deamidated crystallins in the insoluble proteins observed in aged and cataractous lenses. While much is known about the effects of post-translational modifications on homo-oligomers, little is known about the effects on the more relevant hetero-oligomer interactions. We have previously reported that deamidations on the surface of  $\beta$ A3 altered its oligomerization with  $\beta$ B1 dependent on the site [28]. This paper further supports that deamidation can disrupt oligomer formation.

*$\beta$ B1 stabilizes deamidated  $\beta$ A3:*  $\beta$ B1, by forming a hetero-oligomer with DM  $\beta$ A3, prevented the heat induced precipitation of DM  $\beta$ A3. In the absence of  $\beta$ B1, DM  $\beta$ A3 rapidly precipitated at 55 °C. At the concentrations used,  $\beta$ A3 dimers were mixed with  $\beta$ B1 monomers and dimers that then formed hetero-oligomers. Because deamidated  $\beta$ A3 crystallin was not heat stable alone, the increased stability, can be

attributed to the formation of the interactions with  $\beta$ B1 crystallin in the hetero-oligomer.

A stabilizing affect was not seen with  $\beta$ B2, as would be expected since  $\beta$ B2 did not form a hetero-oligomer with DM  $\beta$ A3 at 55 °C. Even though  $\beta$ B2 is more heat stable than DM  $\beta$ A3, it was not able to prevent the precipitation of DM  $\beta$ A3. In contrast, both  $\beta$ B1 and  $\beta$ B2 were able to prevent the thermal denaturation of WT  $\beta$ A3.  $\beta$ B1 appears to be a better “solubilizing partner” than  $\beta$ B2. Because  $\beta$ B1 did not prevent the precipitation of ALDH, it’s stabilizing properties are most likely related to its ability to subunit exchange with the  $\beta$ A3 homo-dimer.

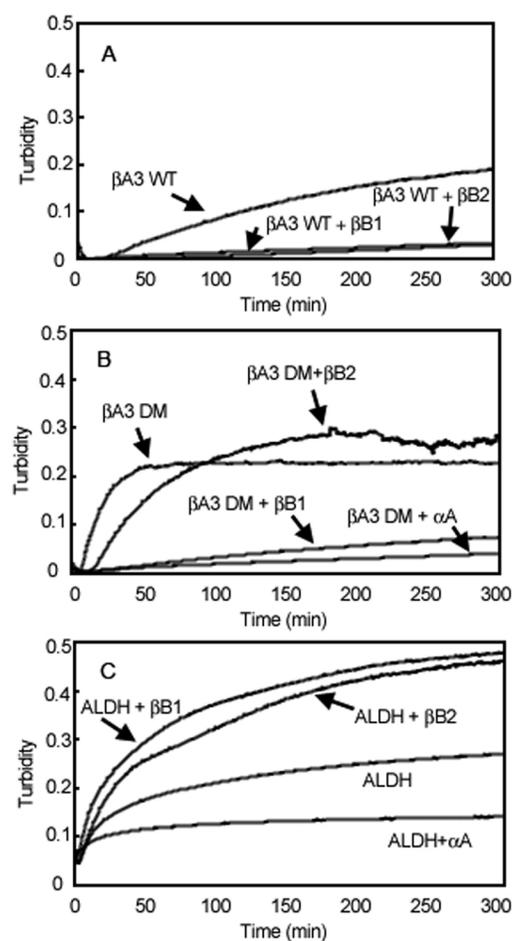


Figure 8. Thermal stability of  $\beta$ -crystallin complexes. Samples were heated at 55 °C and the turbidity of the solutions measured at 405 nm.  $\beta$ B1- and  $\beta$ B2-crystallin were mixed with  $\beta$ A3 WT (A) or  $\beta$ A3 DM (B). As a control,  $\beta$ A3 DM was also mixed with the known chaperone,  $\alpha$ A-crystallin (B). In order to determine if either  $\beta$ B1 or  $\beta$ B2 had chaperone-like activity, the substrate alcohol dehydrogenase (ALDH) was used (C). Both  $\beta$ B1 and  $\beta$ B2 decreased the heat induced precipitation of  $\beta$ A3 WT, but only  $\beta$ B1 decreased the precipitation of  $\beta$ A3 DM. Neither  $\beta$ B1 nor  $\beta$ B2 prevented the precipitation of ALDH.

*Interactions between  $\beta$ -crystallin subunits:* Both WT  $\beta$ A3 and deamidated  $\beta$ A3 formed tetramers with  $\beta$ B1. The molar masses and asymmetric shape of the eluted  $\beta$ B1- $\beta$ A3 hetero-oligomer peak during SEC-MALS suggested a mixture of hetero-tetramers and hetero-dimers, as has previously been reported [22,28]. The  $\beta$ B1-WT  $\beta$ A3 and  $\beta$ B1-DM  $\beta$ A3 hetero-oligomers migrated similarly on BN-PAGE suggesting similar shapes. Far CD and fluorescence analysis also indicated the hetero-oligomers had similar secondary and tertiary structures.

The predicted structures of  $\beta$ B1 and  $\beta$ B2 from crystallography are known and differ [17,18], while the structure of  $\beta$ A3 is not known. Of interest, then, is that the CD spectra of the  $\beta$ B1- $\beta$ A3 hetero-oligomer more closely matched that of  $\beta$ B1 with minima at both 205 and 218 nm. Since, truncated  $\beta$ B1 shows a single minimum at 218 nm [24], the double minima of the  $\beta$ B1- $\beta$ A3 complex reflects the contribution of the long NH<sub>2</sub>-terminal extension of  $\beta$ B1. It cannot be said from these data if the complex also has a bent linker as does the  $\beta$ B1 homo-dimer.

We have previously reported structural changes in DM  $\beta$ A3 [27], which was further supported here by its differing migration from WT  $\beta$ A3 on BN-PAGE. Deamidation by introducing a negative charge at the predicted hydrophobic interface in  $\beta$ A3 disrupted its structure and decreased interaction with  $\beta$ B1 without completely preventing formation of a mixed oligomer. Therefore, deamidation at the predicted interface in the  $\beta$ A3 dimer is also important in the formation of the  $\beta$ A3- $\beta$ B1 hetero-oligomer.

In contrast to  $\beta$ B1,  $\beta$ B2 formed a hetero-dimer with WT  $\beta$ A3 or DM  $\beta$ A3. The hetero-dimer migrated on BN-PAGE as a faint band between the  $\beta$ B2 dimer and  $\beta$ A3 dimer, suggesting the shape was less elongated than the  $\beta$ B2 extended dimer. There was also a slight difference in migration between the  $\beta$ B2-WT  $\beta$ A3 and  $\beta$ B2-DM  $\beta$ A3 dimers, suggesting different shapes of the mixed complexes.

In summary, deamidation decreased formation of hetero-oligomers between  $\beta$ -crystallin subunits with specific  $\beta$ -crystallin interactions protective against insolubilization. An excess accumulation of deamidated  $\beta$ -crystallins in vivo may disrupt normal protein-protein interactions and diminish the stabilizing effects between them, thus, contributing to the accumulation of insoluble  $\beta$ -crystallins during aging and cataracts.

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